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# (54) [Title of the Invention] Nitrogen Trifluoride Exhaust Gas Treatment Method and Apparatus

(57) [Goal] The goal of the present invention is to provide a method and apparatus for NF<sub>3</sub> exhaust gas treatment that has low operational cost, that eliminates NO<sub>x</sub>, and that can effectively prevent equipment corrosion.

[Content of the Invention] Thermal oxidative decomposition apparatus 1 is a specific example of the gas treatment apparatus of the present invention. NH<sub>3</sub> from an NH<sub>3</sub> feed line is added to NF<sub>3</sub>-containing exhaust gas within exhaust gas feed line 5, and the gases are mixed together. Nitrogen flowing from nitrogen feed line 6, together with air flowing from air feed line 7, are fed into gas phase reactor 9. Thermal oxidative decomposable substances within the exhaust gas undergo thermal oxidative decomposition. In particular, NF<sub>3</sub>-derived NO<sub>x</sub> reacts with NH<sub>3</sub> so that the NO<sub>x</sub> is detoxified by formation of nitrogen and water. Residual particles (SiO<sub>2</sub>, etc.), NH<sub>3</sub>, F<sub>2</sub>, SiF<sub>4</sub>, etc. are removed by cooling water supplied by water scrubber 10 of the final stage of reactor section 3. Solids are entrained by the cooling water and are physically removed. Gases that are reactive with water are dissolved, react with NH<sub>3</sub>, are neutralized, and then are discharged through discharge water line 15. Detoxified treatment gas is discharged through exhaust line 16.

## [Limits of the Patent Claims]

[Claim 1] A method for treatment of NF3 exhaust gas,

wherein an exhaust gas that contains at least NF<sub>3</sub> undergoes thermal oxidative decomposition, and the NF<sub>3</sub> oxidative decomposition products and / or derivatives thereof react with NH<sub>3</sub>.

[Claim 2] An apparatus for treatment of NF<sub>3</sub> exhaust gas;

wherein the apparatus comprises a thermal oxidative decomposition apparatus that has a reactor that causes an exhaust gas that contains at least NF<sub>3</sub> to undergo thermal oxidative decomposition; and the reactor reacts NH<sub>3</sub> with NF<sub>3</sub> oxidative decomposition products and / or derivatives thereof.

## [Detailed Explanation of the Invention] [0001]

[Industrial Field of Application] The present invention relates to a method and apparatus for detoxification by cleaning of NF<sub>3</sub> exhaust gas that is generated by the semiconductor, liquid crystal, etc. manufacturing fields.

#### [0002]

[Prior Art] When NF<sub>3</sub> is used for cleaning of semiconductor and liquid crystal manufacturing CVD equipment, as much as 50% of the NF<sub>3</sub> is discharged unreacted.

[0003] Various methods exist for the removal of  $NF_3$ . These  $NF_3$  removal methods include: contacting  $NF_3$  with heated silicon particles, conversion of  $NF_3$  to  $SiF_4$ , followed by adsorption of  $SiF_4$  upon an immobilized alkaline material; contacting  $NF_3$  with heated carbon granules so that  $NF_3$  converts to  $CF_4$ ; and the like.

[0004] Although these methods are effective for the removal of NF<sub>3</sub> alone, these methods have the following deficiencies. The method that uses contact with silicon particles generates  $NO_x$  when the exhaust gas contains oxygen. The method that uses contact with carbon granules discharges  $CF_4$  (a type of Freon compound), and the carbon granules are consumed by reaction with oxygen when the treatment gas contains oxygen.

[0005] Furthermore, each of the methods requires additional treatment to remove toxic components (SiF<sub>4</sub>, NO<sub>x</sub>) within the cleaning exhaust gas, thereby raising costs.

## [0006]

[Problems to be Solved by the Invention] By the use of a thermal oxidative decomposition apparatus that is also effective for treatment of CVD exhaust gas during CVD utilizing SiH<sub>4</sub>, TEOS, etc., NF<sub>3</sub>-containing cleaning exhaust gas treatment can be carried out while reducing equipment cost. When the NF<sub>3</sub> undergoes thermal oxidative decomposition, F<sub>2</sub> and NO<sub>x</sub> are formed by decomposition above 800 °C. Although F<sub>2</sub> and SiF<sub>4</sub> within the cleaning exhaust gas are removed by water scrubbing, NO<sub>x</sub> is discharged in the final exhaust since removal of NO<sub>x</sub> by water scrubbing is inefficient. Problems arise due to generation of NO<sub>x</sub> and the corrosion resulting from acidity of the scrubbing water.

[0007] The goal of the present invention is to provide a method and apparatus for  $NF_3$  exhaust gas treatment that has low operational cost, that eliminates  $NO_x$ , and that can effectively prevent equipment corrosion.

#### [0008]

[Means to Solve the Problems] The present invention comprises the (1) method and the (2) apparatus described below, thereby making possible solution of the above mentioned problems.

- (1) A method for treatment of NF<sub>3</sub> exhaust gas; wherein an exhaust gas that contains at least NF<sub>3</sub> undergoes thermal oxidative decomposition; and the NF<sub>3</sub> oxidative decomposition products and / or derivatives thereof react with NH<sub>3</sub>.
- (2) An apparatus for treatment of NF<sub>3</sub> exhaust gas; wherein the apparatus comprises a thermal oxidative decomposition apparatus that has a reactor that causes an exhaust gas that contains at least NF<sub>3</sub> to undergo thermal oxidative decomposition; and the reactor reacts NH<sub>3</sub> with NF<sub>3</sub> oxidative decomposition products and / or derivatives thereof.

[0009] For the present invention, NF<sub>3</sub>-containing exhaust gas or NF<sub>3</sub> exhaust gas (referred to below simple as "exhaust gas") is a gas that contains NF<sub>3</sub>. Such exhaust gas is taken to be the exhaust gas of an apparatus (such as would be used for semiconductor manufacturing) that utilizes cleaning gas that contains at least NF<sub>3</sub>. Such exhaust gas is taken to mean an exhaust gas that contains process gases and reaction products (derived from reactions between NF<sub>3</sub> and substances to be cleaned within the apparatus during cleaning treatment using cleaning gas). Therefore the NF<sub>3</sub> detoxification treatment of the present invention is taken to include removal of NF<sub>3</sub> derivatives.

[0010] The present invention, by thermal oxidative decomposition of exhaust gas, converts organic components in the process-derived exhaust gas into water and CO<sub>2</sub>, and inorganic components are converted into SiO<sub>2</sub>, etc. metal oxide fine particles. These components are removed by water scrubbing, etc. as desired. At the same time, the NF<sub>3</sub>, etc. to be removed is converted into NO<sub>x</sub>, HF, etc., and this NO<sub>x</sub> and HF are reacted with NH<sub>3</sub>, thereby obtaining a NF<sub>3</sub>-detoxified cleaned gas.

[0011] Per the present invention,  $NH_3$  reacts with the  $NF_3$  thermal oxidative decomposition substances and derivatives thereof. In addition, residual  $NH_3$  can react with substances (generated by the thermal oxidative decomposition treatment and derivatives thereof) that are reactive with  $NH_3$ , such as residual  $SiF_4$  and  $H_2O$  in the cleaning gas.

[0012] Moreover, use of an excessive quantity of NH<sub>3</sub> gas is not a problem since NH<sub>3</sub> is removed by water scrubbing, etc. Since the use of an excess of inexpensive NH<sub>3</sub> gas is harmless, corrosion is prevented, equipment costs decrease, and operational costs can be reduced relative to the prior art (final-stage removal of NO<sub>x</sub>).

[0013] An example that can be suggested of the apparatus that discharges exhaust gas of the present invention is a CVD apparatus. The exhaust gas comprises CVD exhaust gas derived from CVD treatment process gas and / or cleaning exhaust gas during cleaning treatment.

[0014] Examples of the process gas included in such CVD exhaust gas include inorganic raw materials (such as monosilane, disilane, dichlorosilane, etc.) and organic raw materials (such as TEOS (tetraethoxysilane), TMB (trimethyoxyborane), etc.), used either a single substance or as a mixture thereof. The CVD exhaust gas includes such unreacted gases, derivatives thereof, and decomposition reaction products {such as H<sub>2</sub>, CO, alcohols (C<sub>2</sub>H<sub>5</sub>OH, etc.), aldehydes (CH<sub>3</sub>CHO, etc.), and hydrocarbons (C<sub>2</sub>H<sub>4</sub>, etc.)}. The major components are metal oxides (such as SiO<sub>2</sub>, etc.), H<sub>2</sub>O, and CO<sub>2</sub>. Here the meaning of thermal oxidative decomposition includes simple oxidation of non-decomposable compounds (such as hydrogen) and oxidation, etc. of metallic, etc. elements. [TRANSLATOR'S NOTE: The chemical name normally cited as TMB is trimethylborate. "Trimethoxyborane" sounds rather odd.]

[0015] This cleaning exhaust gas comprises at least NF<sub>3</sub>. This cleaning exhaust gas can also comprise reaction products of NF<sub>3</sub> and substances (non-exhausted CVD treatment reagents, etc.) within the CVD apparatus (such as SiF<sub>4</sub>, NF<sub>3</sub> derivatives), cleaning gas, and CVD internal substances physically removed by the cleaning gas.

[0016] Examples that can be suggested of the cleaning gas include  $CF_4$ ,  $C_2F_6$ ,  $SF_6$ ,  $CIF_3$ , etc. For the present invention no particular limitations are placed upon the means used for reacting  $NH_3$  with thermal oxidative decomposition products of  $NF_3$  and / or derivatives thereof. Any method can be used as long as the reaction between  $NH_3$  and reaction product compounds (HF,  $NO_x$ , etc.) is accomplished. Specifically,  $NH_3$  is added and mixed with exhaust gas prior to thermal oxidative decomposition. It is also permissable for  $NH_3$  to also be added simultaneously with the exhaust gas thermal oxidative decomposition reaction and / or thereafter.

[0017] NH<sub>3</sub> may be added simply as NH<sub>3</sub> gas or in another form, such as a gas mixture with air, etc. The quantity of NH<sub>3</sub> used is proportional to the quantity of oxidative decomposition products and / or derivatives thereof. Normally the preferred quantity is at least 4 moles of NH<sub>3</sub> per 1 mole of NF<sub>3</sub> in the exhaust gas.

[0018] No particular limitations are placed upon reaction conditions of the thermal oxidative decomposition treatment of the exhaust gas components and exhaust gas feed conditions, etc. of the present invention as long as such conditions are sufficient for the above mentioned reactions with NH<sub>3</sub>. Thermal oxidative decomposition may be carried out upon those substances within the exhaust gas at least capable of undergoing thermal oxidative decomposition in the presence of oxygen. Therefore, oxygen must exist within the reactor as the exhaust gas is fed to the reactor. Although the method by which oxygen is made to exist within the reactor is arbitrary, a preferred method is to feed oxygen-containing gas (such as air, etc.) together with the normal exhaust gas. Furthermore, an arbitrary gas can co-exist with the exhaust in order to adjust conditions of the thermal oxidative decomposition. For example, an inert gas (nitrogen, etc.) is preferably mixed with the exhaust gas so that the exhaust gas is enveloped by such nitrogen gas that is itself enveloped by oxygen, thereby forming a three-layered gas stream that is fed to the reactor of the thermal oxidative decomposition apparatus. The gas feed lines of the thermal oxidative decomposition apparatus preferably have a three-layered concentric structure. When this type of gas feed line structure is used, NH3 gas may be fed by addition to the normal exhaust gas, or alternatively, by addition to the nitrogen gas or air, etc.

[0019] Furthermore, although the heating means used for thermal oxidative decomposition is arbitrary, a preferred heating means is a temperature-controlled electrical heater which can be provided at the interior wall of a normal reactor. Moreover, the preferred reaction temperature range is 800 to 1000 °C.

[0020] The exhaust gas (referred to below as treated gas), that has undergone thermal oxidative decomposition and / or reaction with NH<sub>3</sub> per the present invention, according to the composition of the treated gas, can be discharged to the environment or can undergo additional arbitrary treatment.

[0021] In particular, a preferred treatment of the treated gas is contact with water, specifically water scrubbing treatment. Metal oxide fine particles ( $SiO_2$ , etc.) generated by the decomposition treatment are captured and removed by this means, and water soluble compounds ( $SiF_4$ ,  $F_2$ , etc.) are removed by dissolution and  $NH_3$  neutralization. Water scrubbing treatment also can cool the treated gas. The type of water scrubbing treatment is arbitrary, although the preferred method is contacting the treated gas with a spray.

[0022] Water scrubbing of the treated gas can be performed by equipment external to the thermal oxidative decomposition apparatus, although this equipment is preferably united (constructed as a single unit) with the thermal oxidative decomposition apparatus as per the present invention. The reactor quite preferably comprises an initial gaseous reactor stage (for thermal oxidative decomposition) and a final water scrubber (for neutralization reaction with NH<sub>3</sub>).

[0023] After water scrubbing treatment of the treated gas, the post-water-scrubbing treated gas can be discharged to the environment or can undergo additional arbitrary treatment, such as known adsorption, etc. treatments, and arbitrary exhaust means can be used (such as an exhaust pipe, etc.). Furthermore, the water scrubbing waste water can be discharged from the system by a discharge line, etc., or alternatively, this waste water can undergo additional treatment. The thermal oxidative decomposition apparatus can be equipped (constructed as a single unit) with the means for discharge of such waste water.

[0024] Even when a large volume of CVD exhaust gas contains NF<sub>3</sub>, this NF<sub>3</sub> can be removed with high efficiency since the thermal oxidative decomposition of exhaust gas of the present invention occurs at high temperature so that treatment can be carried out during a short time period. Moreover, operational costs are lower than the dry adsorption method since treatment is efficiently carried out using electricity (for heating), air, NH<sub>3</sub>, nitrogen, cooling water (including scrubbing water), etc.

[0025] The exhaust gas treatment apparatus of the present invention can be equipped with control devices to appropriately determine treatment conditions during the above mentioned consecutive treatment steps so that the treatment steps are carried out automatically. Such control devices are normally connected to detection devices (such as temperature, pressure, water level, etc. sensors) so that safe and optimum treatment is always carried out.

#### [0026]

[Operation of the Invention]  $NO_x$  and  $F_2$  are generated by oxidative decomposition reactions when  $NF_3$  is fed to the thermal oxidative decomposition apparatus.

[0027]

$$NF_3 + O_2 \rightarrow NO_x + 1.5 F_2$$
 (1)

NO<sub>x</sub> is detoxified by reaction with NH<sub>3</sub> to form N<sub>2</sub> and H<sub>2</sub>O.

$$NO + NH_3 + 0.25 O_2 \rightarrow N_2 + 1.5 H_2O$$
 (2)

$$6 \text{ NO}_2 + 8 \text{ NH}_3 \rightarrow 7 \text{ N}_2 + 12 \text{ H}_2\text{O}$$
 (3)

Moreover, although in the absence of  $NH_3$ , thermal oxidative decomposition of  $NF_3$  generates  $F_2$  then forms the acidic derivative HF upon dissolving in water, neutralization takes place when  $NH_3$  is present.

[0028]

$$F_2 + H_2O \rightarrow 2 HF + 0.5 O_2$$
 (4)

$$HF + NH_3 \rightarrow NH_4F$$
 (5)

Although SiF<sub>4</sub>, which is also contained within the cleaning exhaust gas, generates acidity upon dissolving in water, NH<sub>3</sub> neutralizes this acidity.

[0029]

$$3 \text{ SiF}_4 + 4 \text{ H}_2\text{O} \rightarrow \text{SiO}_2 \cdot 2\text{H}_2\text{O} + 2 \text{ H}_2\text{SiF}_6$$
 (6)

$$2 \text{ NH}_3 + \text{H}_2 \text{SiF}_6 \rightarrow (\text{NH}_4)_2 \text{SiF}_6$$
 (7)

Therefore the scrubbing water becomes acidic if the number of moles of  $NH_3$  is not at least 4 times that of  $NF_3$  and at least 4/3 times that of  $SiF_4$ .

[0030] Furthermore, although reactions (1), (2), and (3) are extremely rapid above 900 °C, NH<sub>3</sub> itself undergoes partial oxidation above 1000 °C. Therefore treatment above 900 °C and below 1000 °C is effective.

#### [0031]

[Working Examples] Although the present invention is explained below by the use of working examples, the present invention is not limited to these working examples. As shown in Figure 1, a specific example of the exhaust gas treatment apparatus of the present invention is thermal oxidative decomposition apparatus 1. The major components of thermal oxidative decomposition apparatus 1 are exhaust gas feed section 2, reactor section 3, and discharge section 4.

[0032] Exhaust gas feed section 2 comprises exhaust gas feed line 5, nitrogen feed line 6, and air feed line 7. These feed lines form a three-layer concentric structure. NH<sub>3</sub> feed line 8 is connected to exhaust gas feed line 5.

[0033] Reactor section 3 comprises gas phase reactor 9 and water scrubber 10. Ceramic heater 11 is provided at the exterior wall as a heat source in order to carry out thermal oxidative decomposition of the exhaust gas within the mixed gas fed from the gas feed section, and reactor section 3 is equipped with thermocouples 12 and 13. Water scrubber 10 is provided as a following stage after gas phase reactor 9. Cooling water 14 is supplied to water scrubber 10.

[0034] Discharge section 4 comprises discharge water line 15 (discharges scrubbing waste water) and exhaust line 16 (vents treated gas). Item 17 is a gas sampling line, and item 18 is a gas sampling line. The exhaust gas treatment of the present invention is carried out as described below using the above mentioned apparatus.

[0035] NH<sub>3</sub> is added via the NH<sub>3</sub> feed line to the NF<sub>3</sub>-containing exhaust gas within exhaust gas feed line 5. Nitrogen flows from nitrogen feed line 6, and air flows from air feed line 7. These gas feeds are released together into gas phase reactor 9. Substances within the exhaust gas that are capable of thermal oxidative decomposition undergo thermal oxidative decomposition. In particular, NO<sub>x</sub> derived from NF<sub>3</sub> reacts with NH<sub>3</sub> to form nitrogen and water so that NO<sub>x</sub> is detoxified. Remnant SiO<sub>2</sub>, etc. fine particles, NH<sub>3</sub>, F<sub>2</sub>, SiF<sub>4</sub>, etc. are removed by cooling water in the water scrubber 10 that follows reactor section 3. Solids are removed by physical entrapment. Water soluble gases are removed by dissolution, followed by neutralization by reaction with NH<sub>3</sub>. These removed materials are discharged through discharge water line 15. Furthermore, the detoxified treated gas is discharged through exhaust line 16.

## [0036] Working Example 1

A treatment experiment was carried out using the thermal oxidative decomposition apparatus shown in Figure 1. Exhaust gas (NF<sub>3</sub> diluted to a concentration of 5,000 ppm by  $N_2$  gas) was fed at a rate of 20 liters per minute to the exhaust gas feed line. NH<sub>3</sub> was added at rates that were increased step-wise. Treatment results are shown in Table 1. Moreover, other feeds to the thermal oxidative decomposition apparatus were as follows: air flow was 10 liters per minute, nitrogen flow was 10 liters per minute, and cooling water flow was 4 liters per minute. The gas phase reactor temperature was 950 °C. The scrubbing waste water was slightly alkaline: at least 20,000 ppm NH<sub>3</sub> concentration.

[0037] [Table 1]

Feed Ga	s Stream	Discharge Gas					
NF <sub>3</sub> (ppm) NH <sub>3</sub> (ppm)		NF <sub>3</sub> (ppm)	NH <sub>3</sub> (ppm)	NO <sub>x</sub> (ppm)	F <sub>2</sub> (ppm)		
5,000	0	40		2,300	< 1		
5,000	5,000	30	< 1	1,500	< 1		
4,900	10,000	22	< 1	480	< 1		
4,900	15,000	13	< 1	75	< 1		
4,900	20,000	3	< 1	2	< 1		
4,900	25,000	. 7	< 1	1	<1		

## [0038] Working Example 2

TEOS was used as a CVD process gas. NF<sub>3</sub> was used as a cleaning gas. Results of treatment of exhaust gas from a single-wafer type CVD apparatus are shown in **Table 2**.

[0039] Exhaust gas flow rate was 20 liters per minute so that the  $NH_3$  concentration was 20,000 ppm. Other treatment conditions were that same as those of Working Example 1. The scrubbing water pH range was 7.0 to 10.5.

[0040] [Table 2]

(units = ppm)

No.	NF <sub>3</sub>		SiF₄		F <sub>2</sub>		C <sub>2</sub> H <sub>5</sub> OH		CH₃CHO		СО		NH,
	in	out	in	out	in	out	in	out	in	out	in	out	out
1	3,100	8	910	< 1	800	< 1	180	< 10	85	< 10	1,500	620	< 1
2	500	4	82	< 1	< 1	< 1	2,300	< 10	100	< 10	6,700	2,400	< 1
3	4	< 1	95	< 1	< 1	< 1	1,200	< 10	330	36	2,900	1,050	<1
4	5,300	10	410	< 1	240	< 1	300	< 10	150	15	2,300	750	< 1
5	1,100	5	56	< 1	2	< 1	45	< 10	13	< 10	200	30	< 1
6	7,200	25	200	< 1	50	< 1	340	< 10	160	20	3,300	1,200	< I

#### [0041]

[Results of the Invention] Although CVD exhaust gas and cleaning exhaust gas used by a single-wafer type CVD apparatus are frequently discharged without complete separation from one another and without discharge, the present invention always performs good treatment since both types of exhaust gas (CVD exhaust gas and cleaning exhaust gas) are treated in the identical apparatus. [TRANSLATOR'S NOTE: The phrase "without discharge" appears to be an error in the source text.]

[0042] Concentration of NF<sub>3</sub> in the exhaust gas of a single-wafer type apparatus varies greatly. Per the method of the present invention, NH<sub>3</sub> must always be added in a manner corresponding to the maximum concentration. Although NH<sub>3</sub> is sometimes in excess, excess NH<sub>3</sub> is removed and adsorbed by water scrubbing so that exhaust gas of the thermal oxidative decomposition apparatus does not contain NH<sub>3</sub>. Furthermore, although the scrubbing solution becomes weakly alkaline, corrosion is not a problem.

## [Simple Explanation of the Illustrations]

[Figure 1] This drawing is used to explain a working example of the exhaust gas treatment apparatus of the present invention.

## [Explanation of Items]

1	••••	Thermal oxidative decomposition apparatus
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2 .... Exhaust gas feed section

3 .... Reactor section

4 .... Discharge section

5 .... Exhaust gas feed line

6 .... Nitrogen feed line

7 .... Air feed line

8 .... NH<sub>3</sub> feed line

9 .... Gas phase reactor

10 .... Water scrubber

11 .... Ceramic heater

12 .... Thermocouple

13 .... Thermocouple

14 .... Cooling water

15 .... Discharge water line

16 .... Exhaust line

17 .... Gas sampling line

18 .... Gas sampling line

